

# Review Sheet

## Types of problems

Stoichiometry

Simple road map

Limiting reactant

Precipitate

Acid/base

Empirical Formula

Mass spec (%)

Combustion analysis

Molecular formula

Gas Laws

Ideal gas law

Combined gas law (and 3 gas laws)

Graham's Law

Avogadro's Law

Thermochemistry

Energy/heat/work

Heat transfer (final temp, specific heat)

Hess' Law

Enthalpies of formation

## Tools

molecular weight

% composition

grams  $\Leftrightarrow$  moles

Polyatomic ions

ionic/covalent/acid nomenclature

Solubility table

Pressure/temp conversions

## Concepts

Kinetic/molecular theory

Atomic theory (atomic structure)

## Random stuff

STP, standard conditions

22.4 L/mol

273 K

760 torr

$\text{Ag}^+$ ,  $\text{Zn}^{2+}$